

Al-C-Si (Aluminum-Carbon-Silicon)

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The previous experimental work on this system includes the efforts of [1987Ode], [1988Lee] and [1990Via]. The compilation by [1995Vil] presented a liquidus projection, isothermal sections at 2150, 2000, 900, 572, 557, and 500 °C and vertical sections along the Al₄C₃-SiC and Al-SiC joins. [1996Gro] reported a thermodynamic assessment of this system and computed isothermal sections at 2150 and 2000 °C, a vertical section along Al₄C₃-SiC join and a liquidus projection.

Binary Systems

The Al-C system depicts the stoichiometric compound Al₄C₃ (*D*₇₁-type rhombohedral). The Al-Si phase diagram is a simple eutectic system with the eutectic reaction at 577 °C and 12.2 at.% Si. The C-Si phase diagram [Massalski2] depicts a stoichiometric phase SiC. The stable modification is βSiC (*B*₃, sphalerite-type cubic).

Ternary Compounds

Two ternary compounds along the Al₄C₃-SiC join have been reported at Al₄C₃·SiC (Al₅C₃N-type hexagonal;

$a = 0.32771$ nm and $c = 2.1676$ nm, denoted τ_1 here) and at 2Al₄C₃·SiC (τ_2 ; *hP16*, $a = 0.33127$ nm and $c = 1.9242$ nm). The third reported phase at Al₄C₃·2SiC has not been confirmed. Also, no polymorphic transformation in Al₄C₃·SiC has been confirmed.

Computed Phase Equilibria

In their thermodynamic description of this system, [1996Gro] modeled the liquid as a solution phase, using the interaction parameters for the liquid from the binary systems. The small solubilities of the other components in (Al), (Si) and Al₄C₃ were taken into account. The ternary compounds Al₄C₃·SiC and 2Al₄C₃·SiC and SiC were treated as stoichiometric compounds and their Gibbs energy descriptions were obtained from heat capacity data. Two isothermal sections at 2150 and 2000 °C, a liquidus projection, and a vertical section along the Al₄C₃-SiC join were computed by [1996Gro]. The two isothermal sections are compared with the experimental data of [1987Ode] in Fig. 1 and 2. The agreement is satisfactory.

Recently, [2000Aks] demonstrated the use of the Al-C-Si phase equilibria below 1000 °C to the microstructural analysis of metal matrix composites containing particles or fibers of SiC.

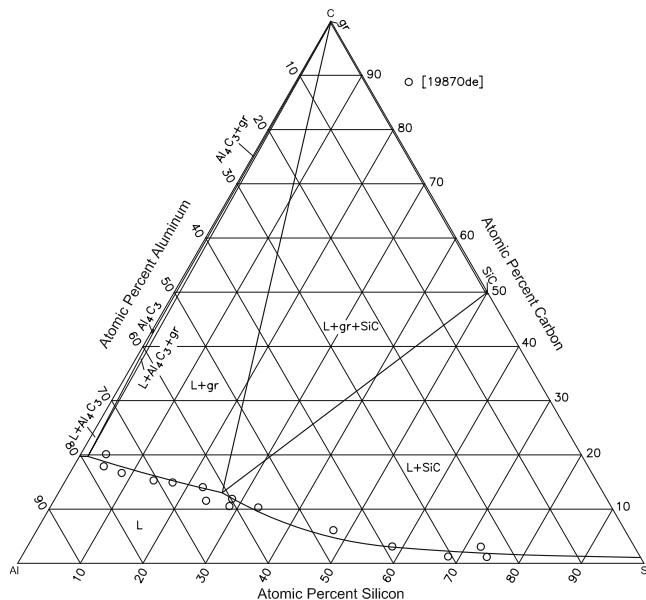


Fig. 1 Al-C-Si computed isothermal section at 2150 °C [1996Gro]

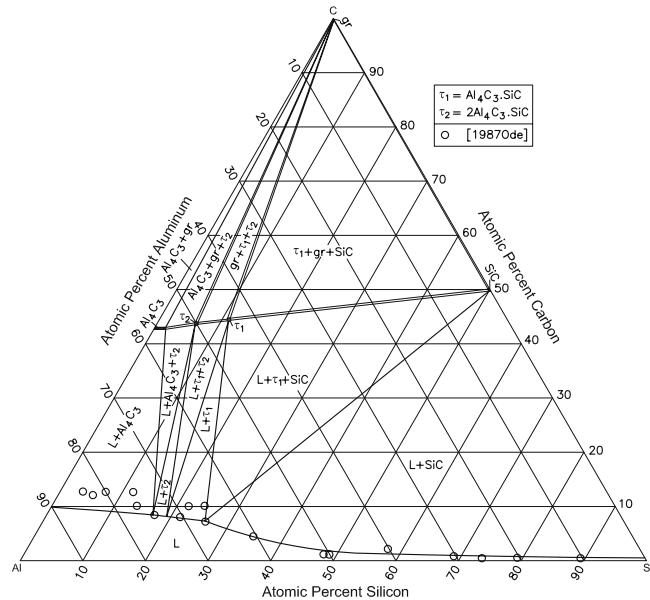


Fig. 2 Al-C-Si computed isothermal section at 2000 °C [1996Gro]

Section II: Phase Diagram Evaluations

References

- 1987Ode:** L.L. Oden and R.A. McCune, Phase Equilibria in the Al-Si-C System, *Metall. Trans. A*, 1987, **18**, p 2005-2014
- 1988Lee:** D.J. Lee, M.D. Vaudin, C.A. Handwerker, and U.R. Kattner, Phase Stability and Interface Reactions in the Al-Si-C System, *Mater. Res. Soc. Symp. Proc.*, 1988, **120**, p 357-365
- 1990Via:** J.C. Viala, P. Fortier, and J. Bouix, Stable and Metastable Phase Equilibria in the Chemical Interaction Between Aluminum and Silicon Carbide, *J. Mater. Sci.*, 1990, **25**, p 1842-1850
- 1995Vil:** P. Villars, A. Prince, and H. Okamoto, Al-C-Si, *Handbook of Ternary Alloy Phase Diagrams*, ASM International, Materials Park, OH, 1995, **3**, p 2893-2900
- 1996Gro:** J. Grobner, H.L. Lukas, and F. Aldinger, Thermodynamic Calculation of the Ternary System Al-Si-C, *CALPHAD*, 1996, **20**(2), p 247-254
- 2000Aks:** A.A. Aksenov, N.A. Belov, and S.V. Medvedeva, The Al-Si-C Phase Diagram and Its Use for Microstructural Analysis of MMC_p and MMC_f Composite Materials, *Z. Metallkd.*, 2000, **92**(9), p 1103-1110